





Parameters, Sample Points and Cost of Installation

• Many factors to consider depending on your water characteristics and process equipment

AppliedSpectrometry

Associates. Inc

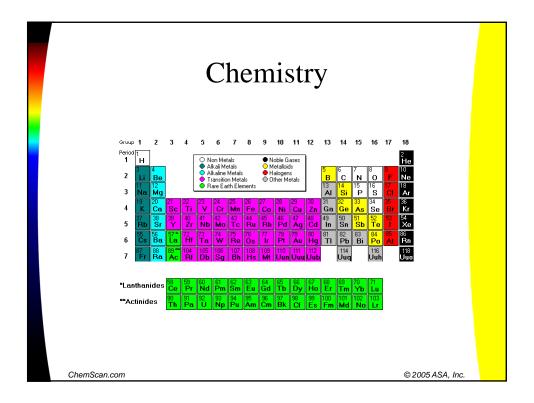
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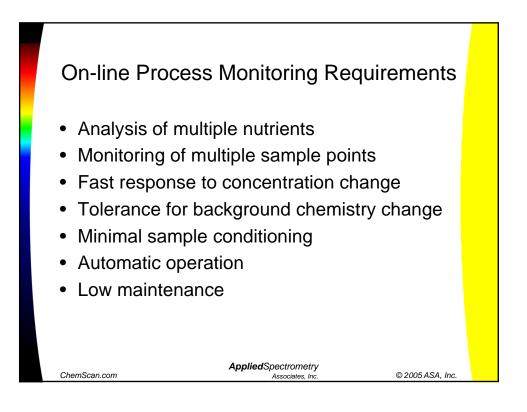
- Where to test
- What to test

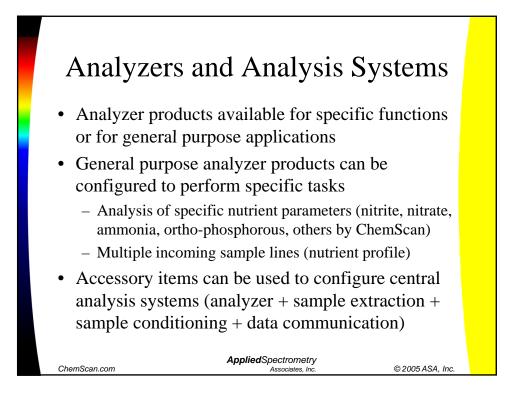
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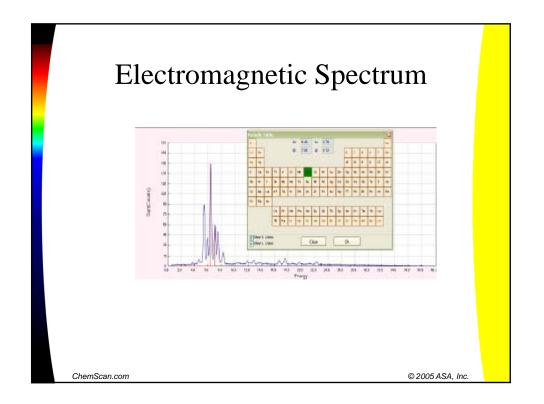
- How often to test
- How much to operate and maintain
- How much does it cost to install

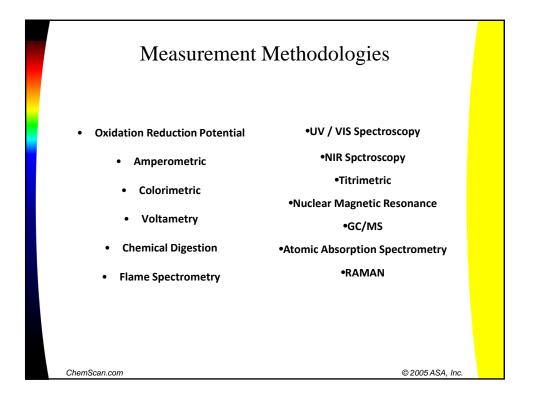
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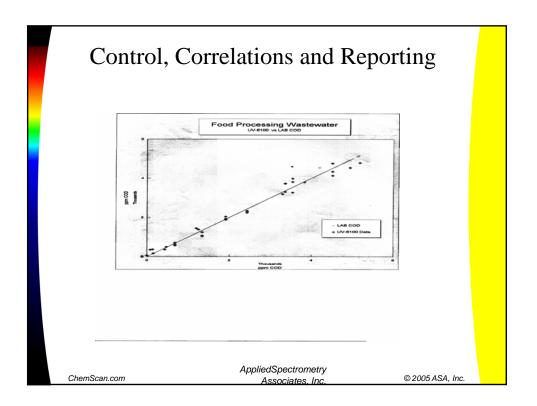


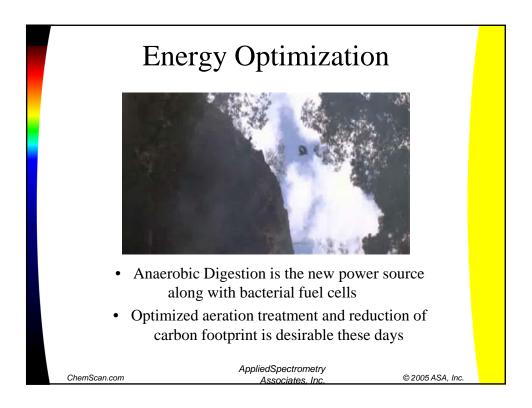


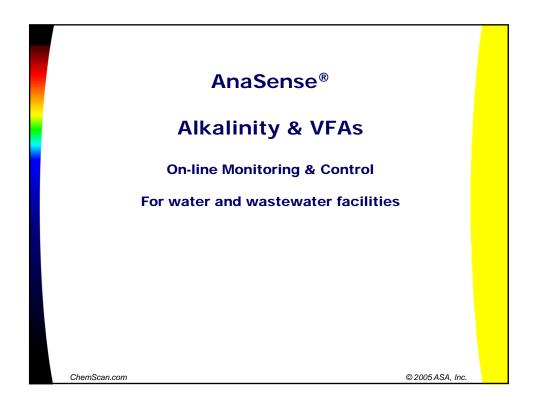


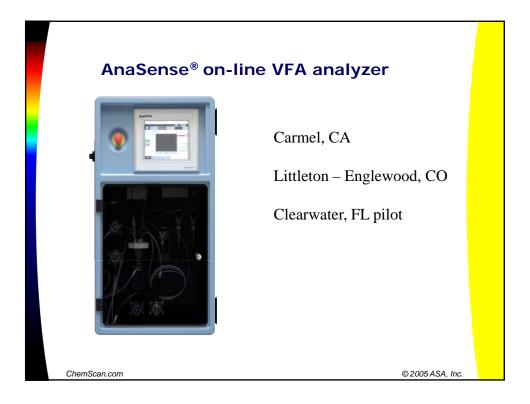


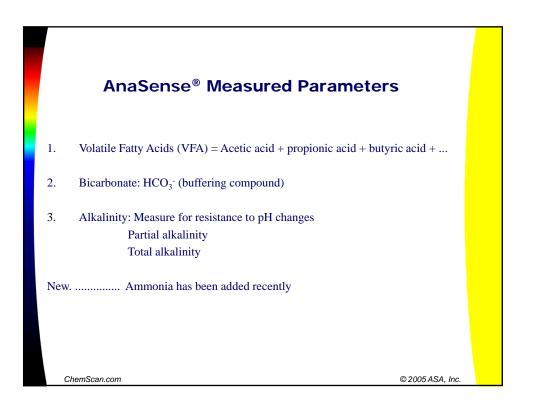


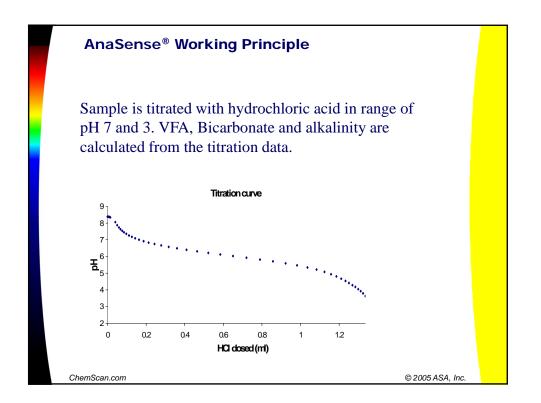


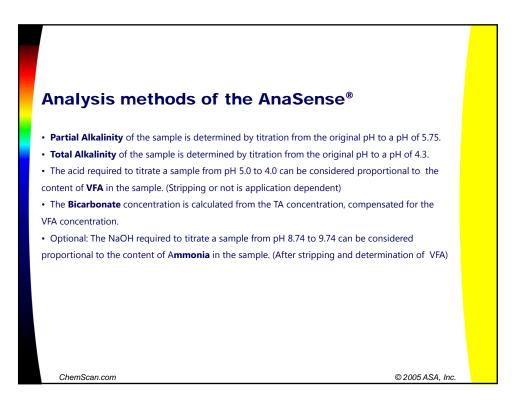


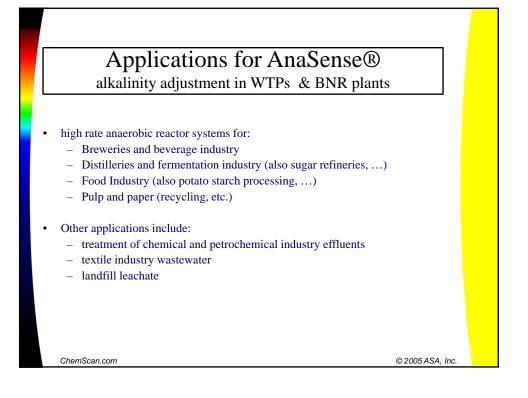


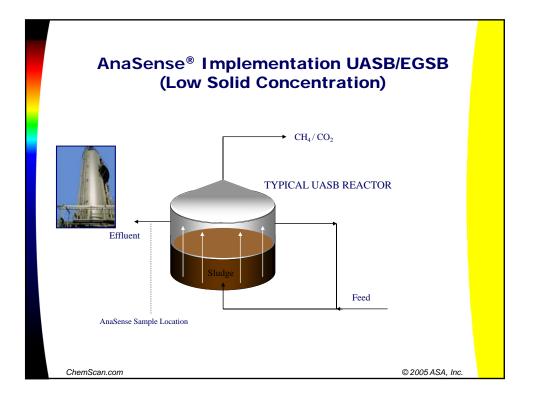


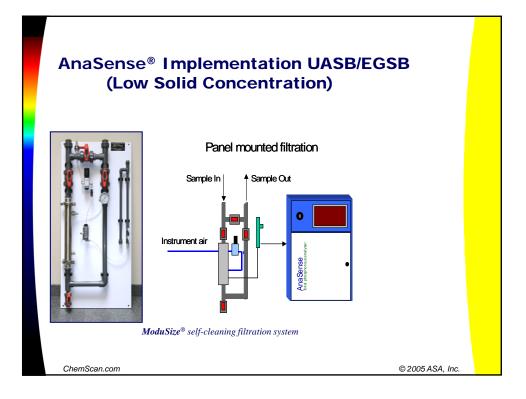


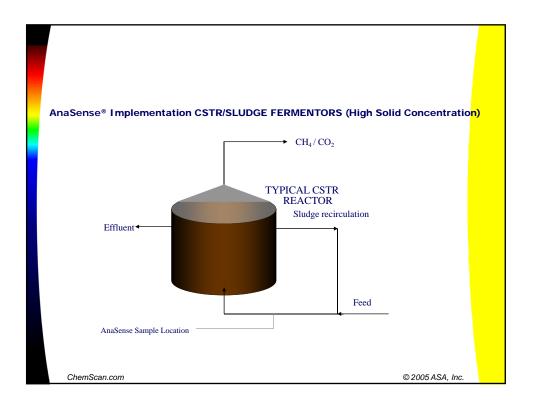


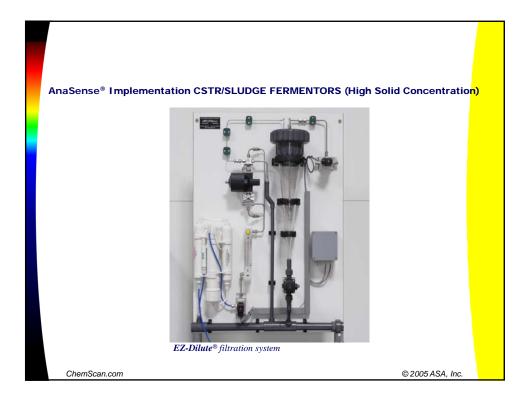


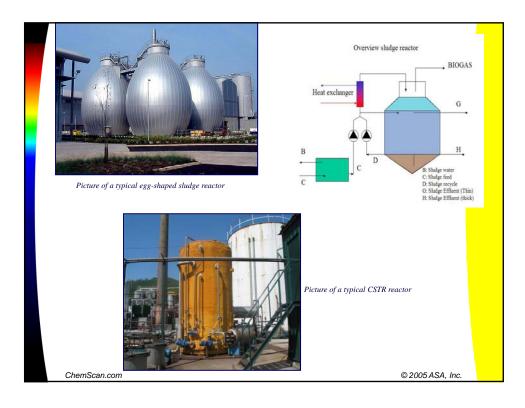


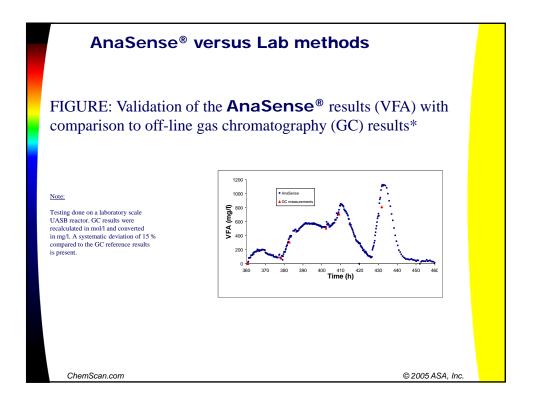


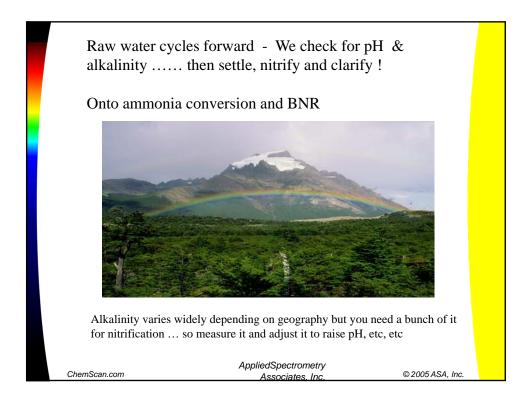










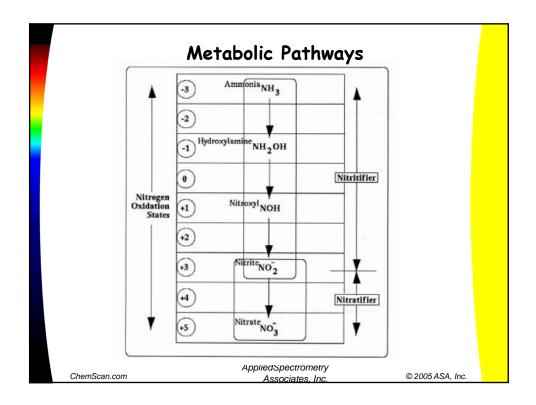


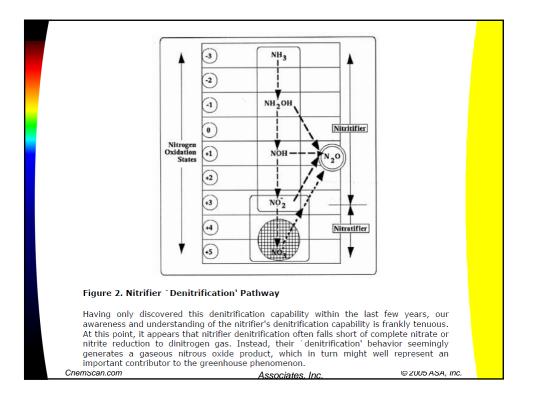
Bacterial Groupings and Genetics

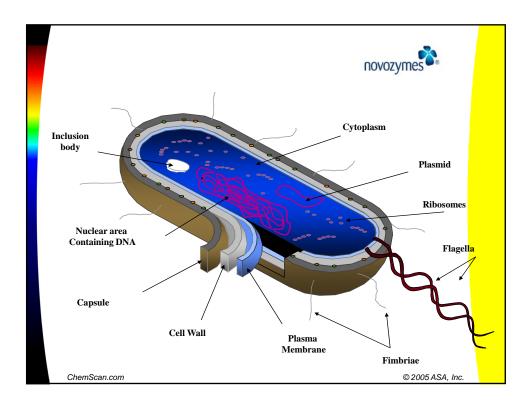
The nitrifying bacterial clan includes two distinctly different subsets, based on their consumption of either ammonia or nitrite. Table 1 provides a breakdown of these affiliated bacterial genus and species members, covering a total of eight separate nitrifying bacteria, including: five species of ammonia-oxidizers (often called `*nitritifiers*'), and three nitrite-oxidizers (`*nitratifiers*').

TABLE 1. Taxonomy of Nitrifying Bacteria

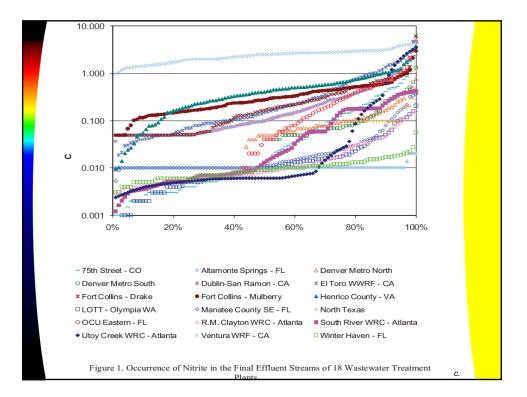
Taxonomic Group			ne S	oil		otility Pote Non-M			
Nitritifiers Nitrosomonas e Nitrospira briens Nitrosococcus o Nitrosococcus o Nitrosolobus mu	sis * itrosus ceanus	* * *	*	*	* * * *	* * *			
Nitratifiers Nitrobacter wind	ogradsky	*	*	*	possibl	y in * continu culture	ious		
Nitrospina graci Nitrococcus mol		*			*	culture			
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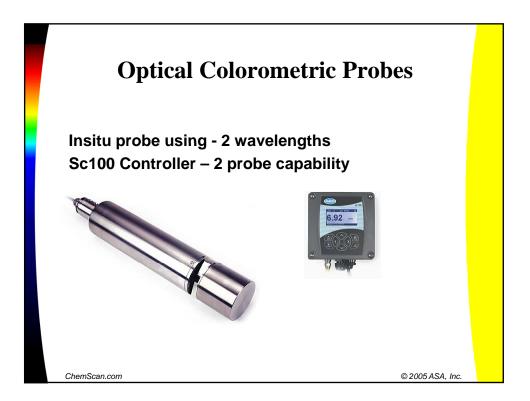


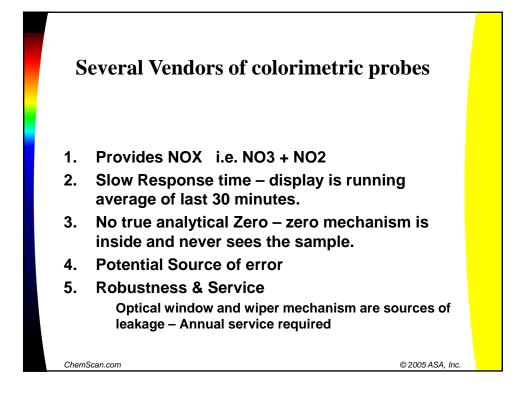


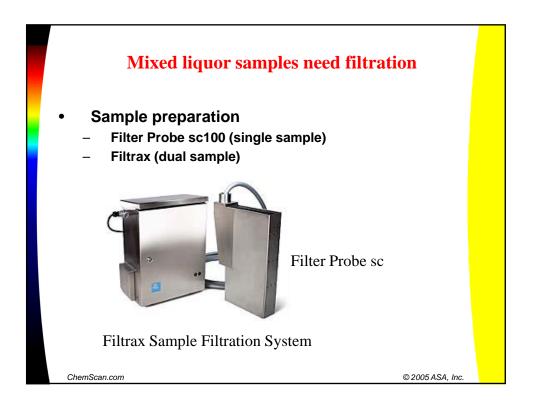


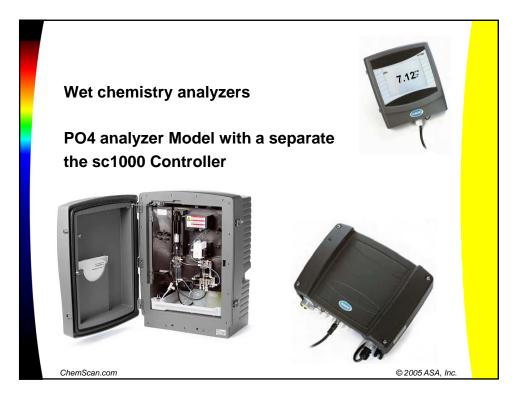


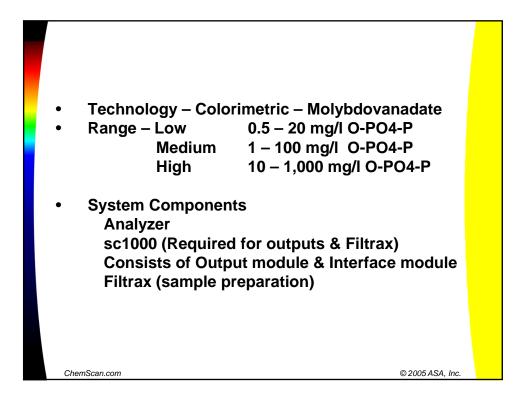
Reduction of Nitrite & Nitrate:	
The nitrate reducing bacteria are facultative anaerobic heterotophs.	
Therefore, an organic carbon source is required.	
For the following equations methanol has been used as the carbon source.	
First Energy Reaction: $6 \text{ NO}_3^{-} + 2 \text{ CH}_3\text{OH} \Rightarrow 6 \text{ NO}_2^{-} + 2 \text{ CO}_2^{-} + 4 \text{ H}_2\text{O}$ nitrate methanol nitrite	
Second Energy Reaction: $6 \text{ NO}_2 + 3 \text{ CH}_3\text{OH} \Rightarrow 3 \text{ N}_2 + 3 \text{ CO}_2 + 3 \text{ H}_2\text{O} + 6 \text{ OH}^2$ nitrite methanol nitrogen gas	
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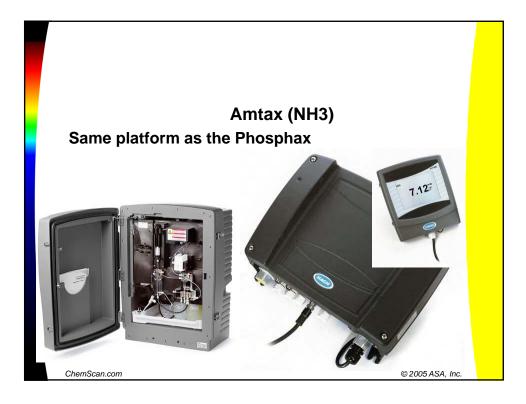


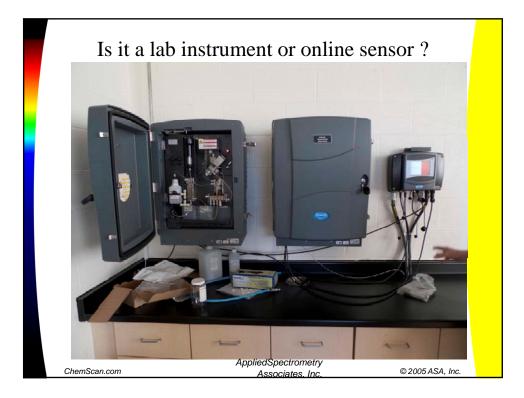


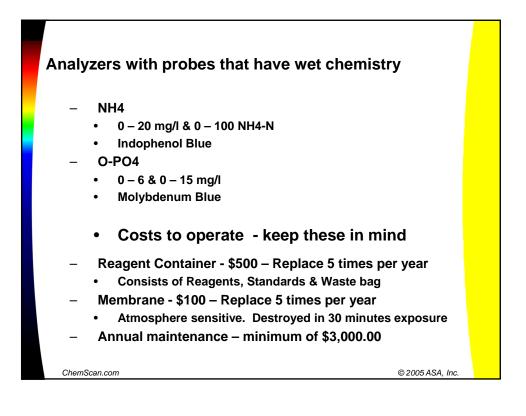


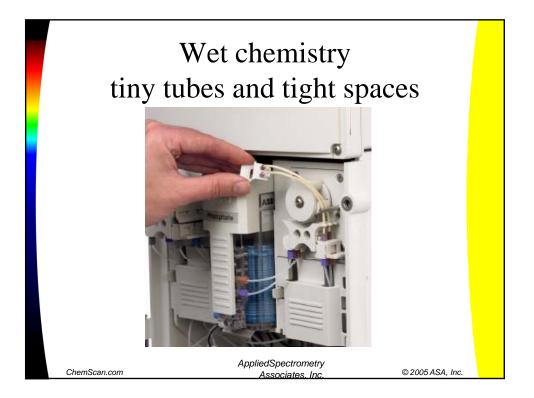












Blue or yellow method Colorimetric or spectrometric

TresCon® OP 210

- Yellow method
- Continuous background compensation
- Continuous/Discontinuous
 operation selectable

On-line orthophosphate measurement

- Control or feedback control of chemical phosphate precipitation, e.g. precipitating agent addition with simultaneous precipitation
- Monitoring biological phosphate elimination
- Measuring the phosphate pollution in natural waters
- Monitoring the phosphate concentration in the drinking water

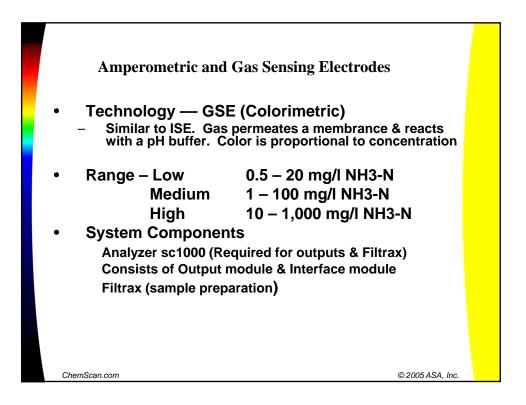
Measuring Principle

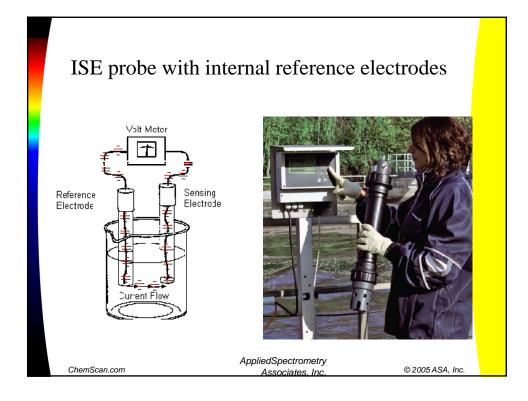
The PO_4 module uses the vanadate/molybdate method (yellow method) for determining the orthophosphate content. A reagent reacts with phosphate in the sample to color the sample solution yellow. The intensity of this color is recorded photometrically and evaluated as a measure of the phosphate content.

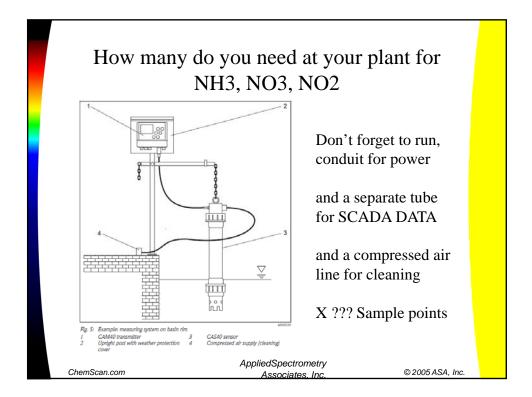
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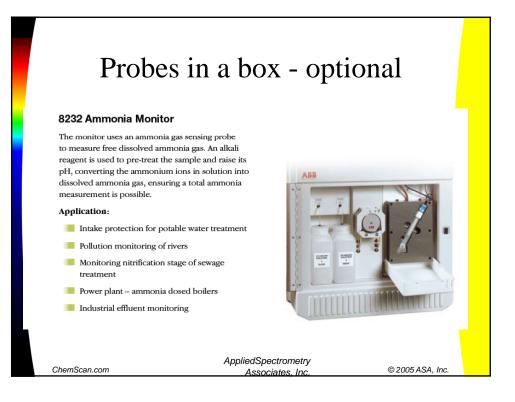
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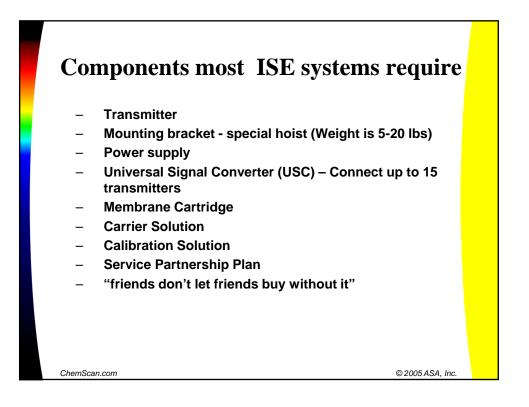
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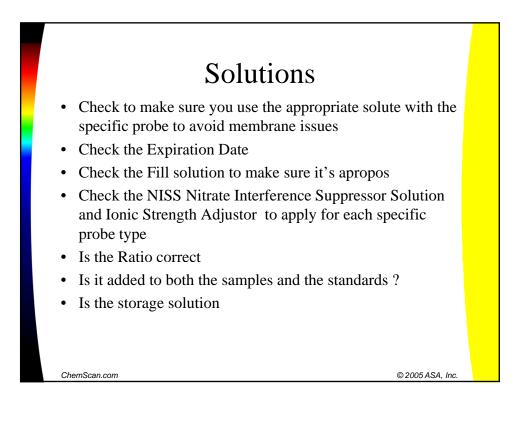


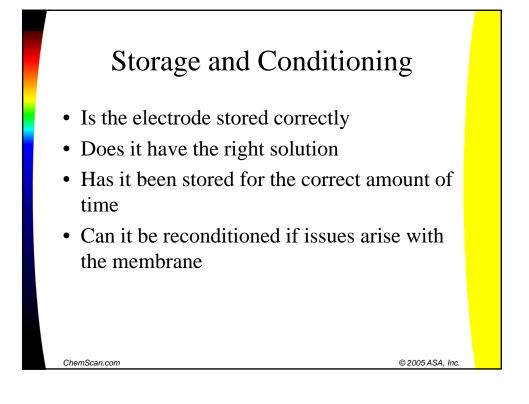


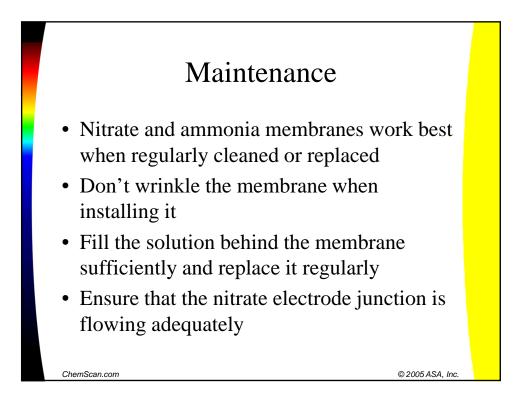


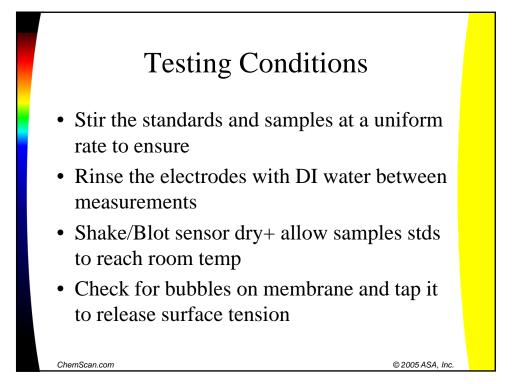


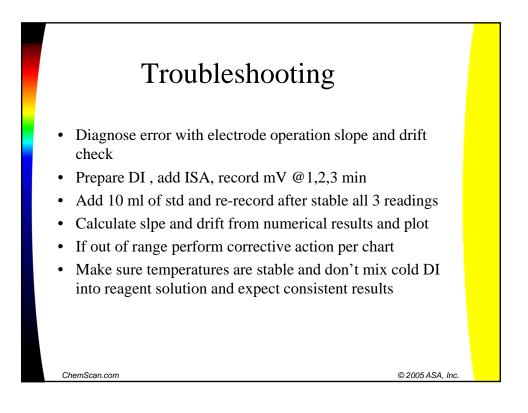
ISE & GSE Probe O&M	
Article by Thermo Fisher Scientific	
Gayle Gleischauf – Applications Manager	
Recommendations for Optimizing Performance	
Follow Recommended Manufacturer Storage and Maintenance Procedures	
published in The Georgia Operator - Summer 2010	
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San Jo	se - Inde	ependent	studies
		pendente	

I able 4. Ammonia meters results in the mixed liquor.								
Brand - S	Sensor	WTW - Varion [®] Plus 700 IQ				Hach - NH4Dsc		
Location		1 2 3				1	2	
Nber of s	amples	44	9	8	12	12	40	
Av. Conc. ⁽¹⁾	Lab.	7.0 (±1.5)	4.7 (±1.5)	1.9 (±0.8)	0.2 (±0.1)	20.2 (±2.7)	5.9 (±2.0)	
(mg/L)	Meter	7.3 (±1.5)	4.5 (±1.5)	1.9 (±0.9)	0.4 (±0.1)	20.9 (±6.9)	7.6 (±3.6)	
Differenc	e ⁽²⁾	-0.29 mg/L -5.0%	0.20 mg/L 3.2%	-0.02 mg/L -5.0%	0.23 mg/L 53.1%	-0.65 mg/L -2.4%	-1.66 mg/L -55.7%	
Conclusio	on	Recom.	Recom.	Recom.	Not Recom.	Not Recom.	Not Recom.	

Table 4. Ammonia meters results in the mixed liquor

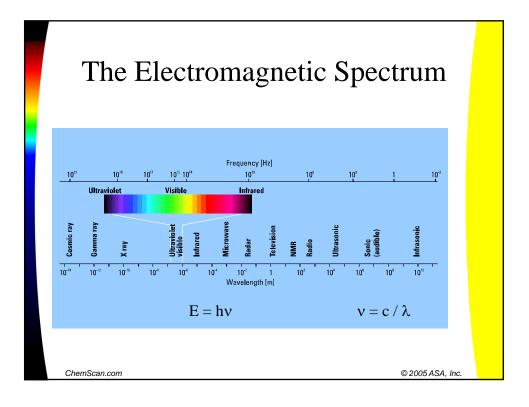
Concentration in terms of NH₄-N - Standard deviation is specified in bracket.
 Difference (mg/L) is between laboratory measurements and meter readings (mg/L).

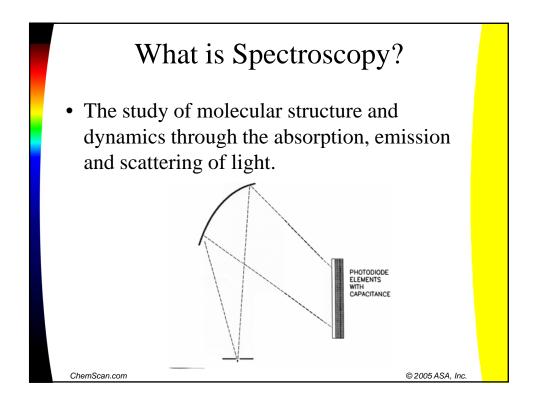
Difference (%) is the ratio between the difference (mg/L) and the laboratory measurements. Av. Conc. = Average Concentration.

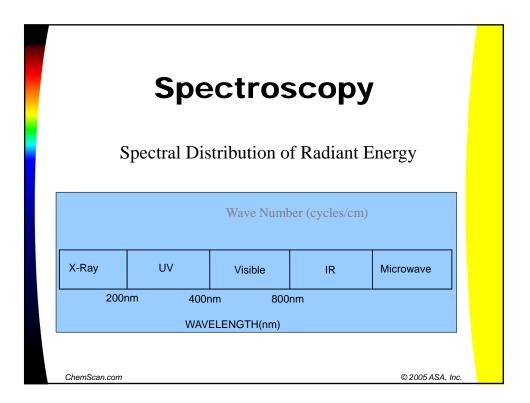
When the NH₄ concentration is below 1.9(\pm 0.8) mgNH₄-N/L (Location 4), discrepancy of 53% (0.23 mg/L) was observed. The meter is not accurate enough to detect precisely low ammonia concentration (Figure 7-d); the limit of this on-line meter was reached.

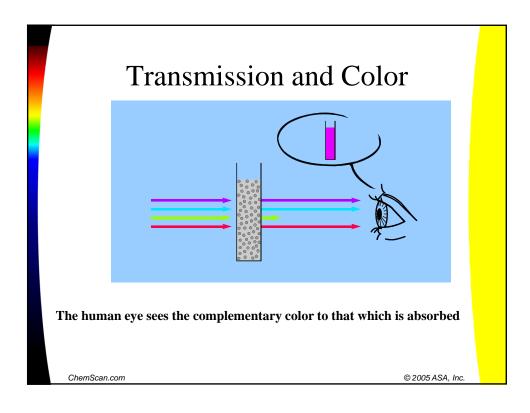
The WTW ammonia meter based on ISE measuring principle is only recommended for application with NH_4 concentration in the mixed liquor above 2 mg/L.

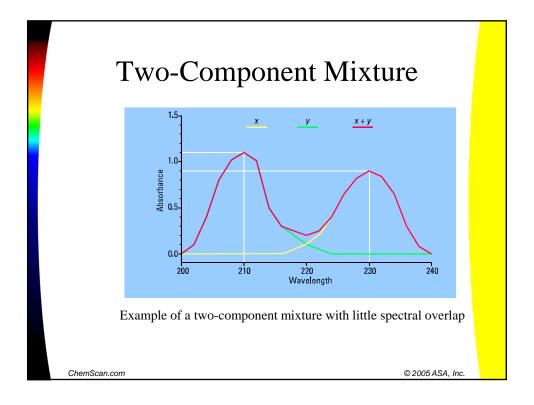
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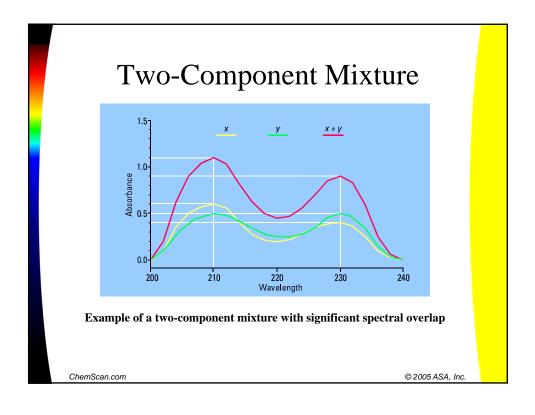


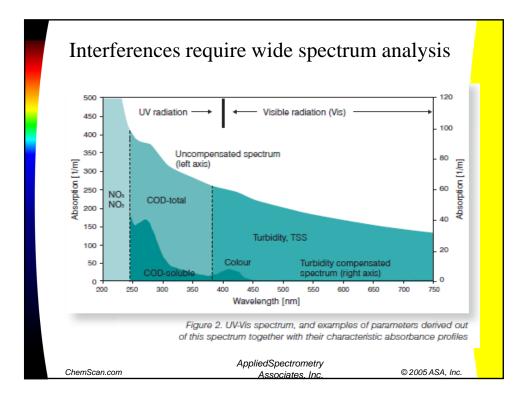


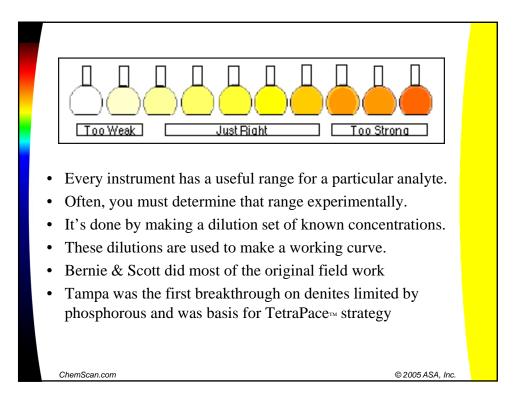




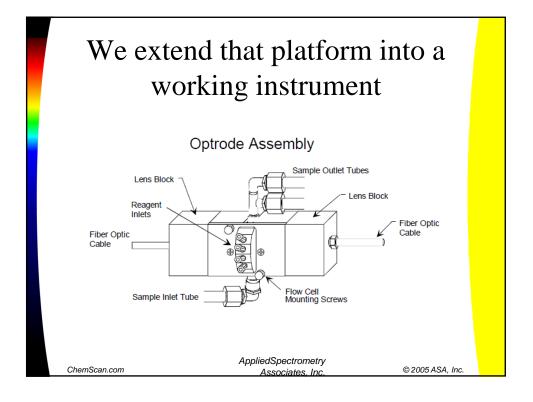


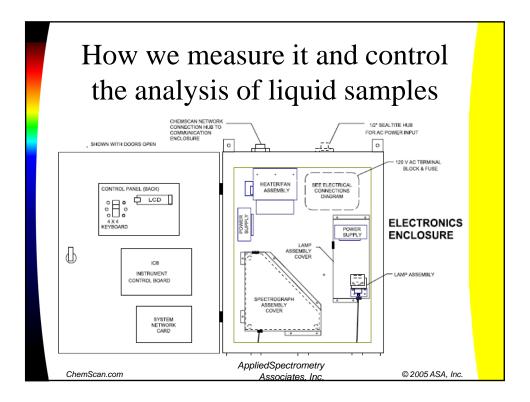


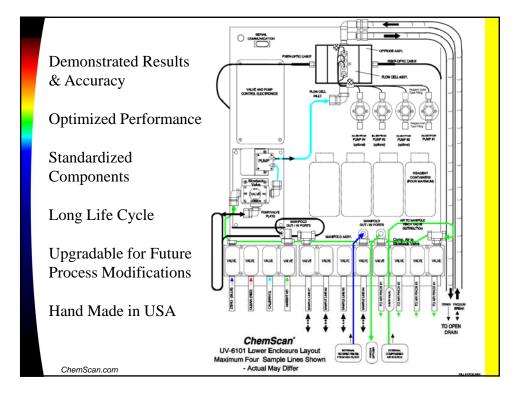


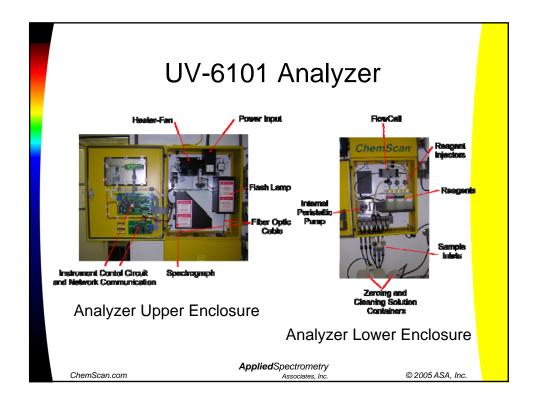


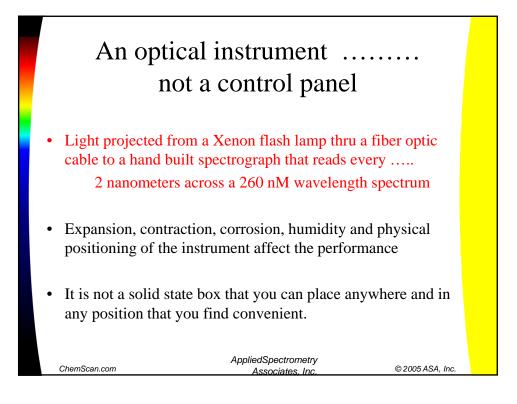


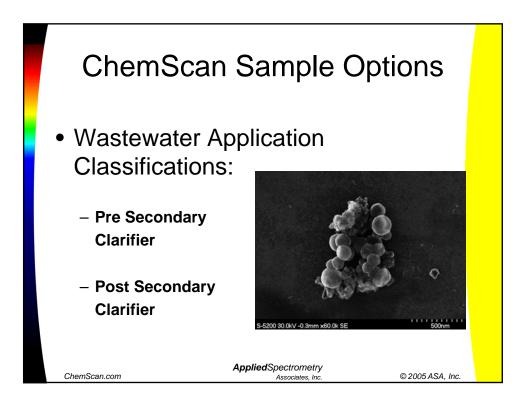


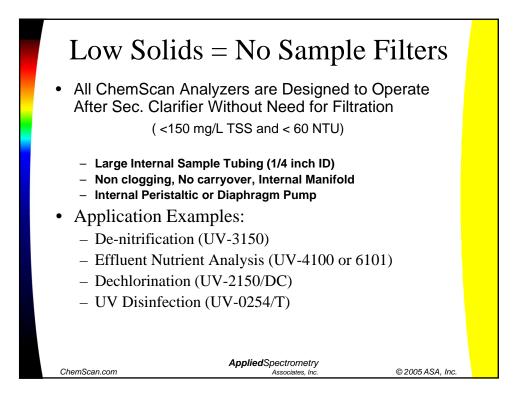


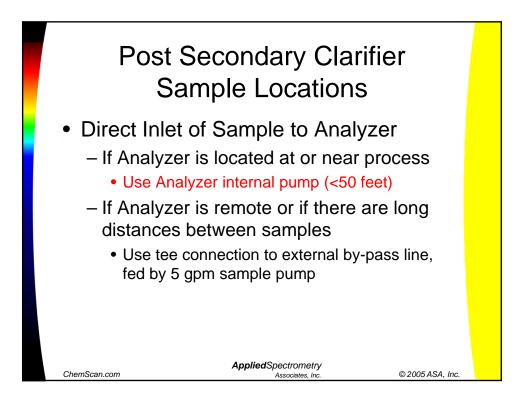


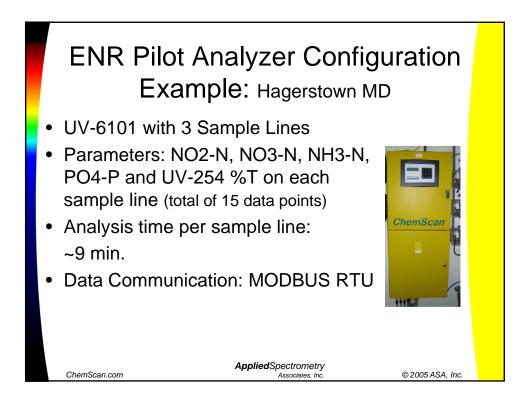


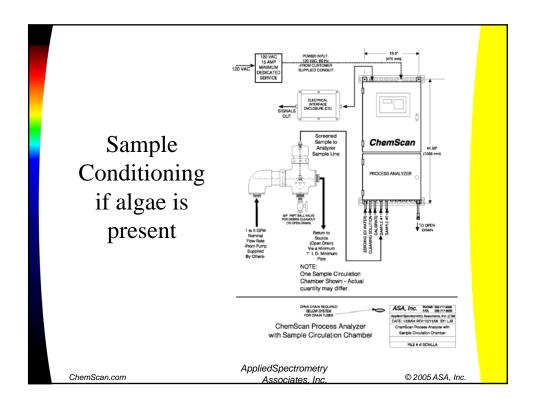


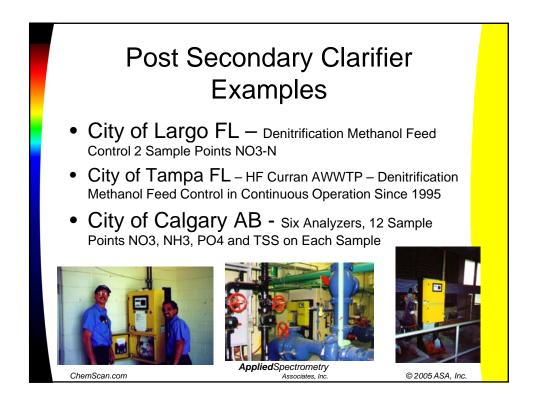


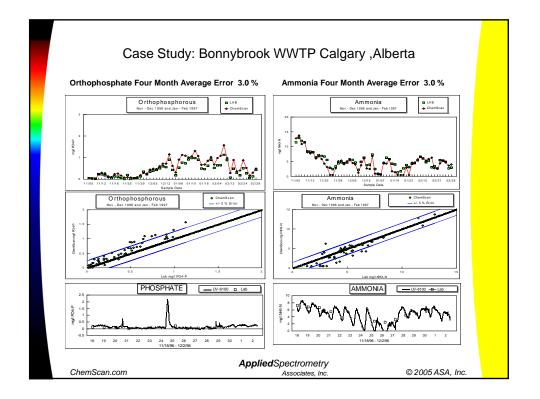


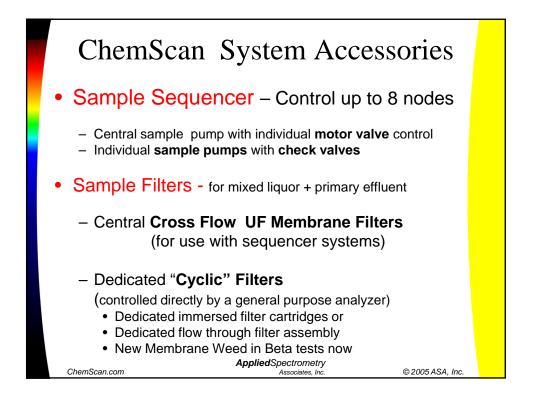


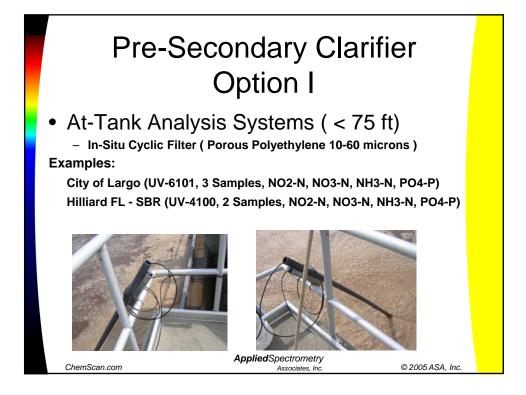


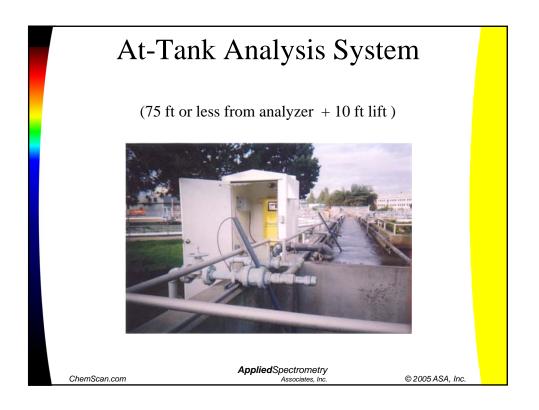


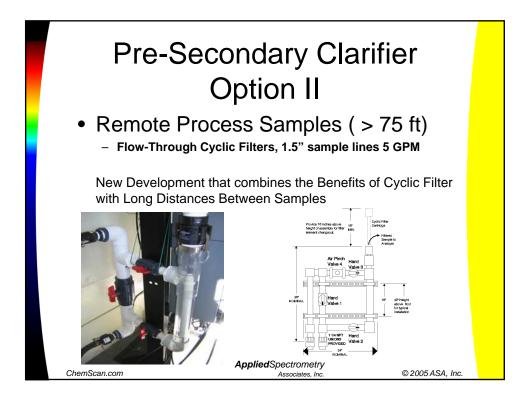


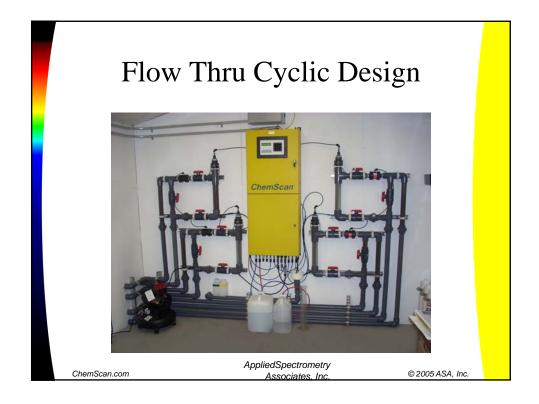


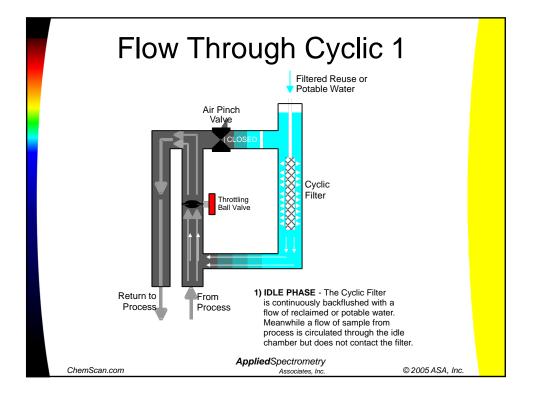


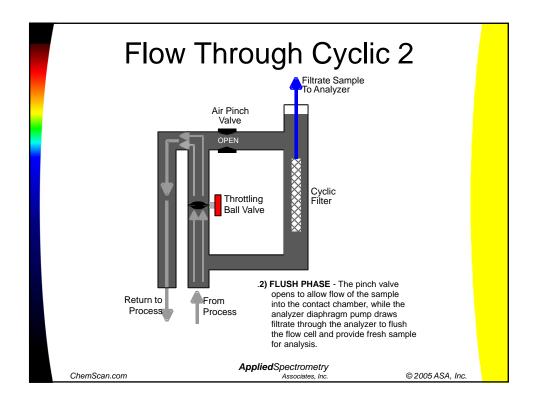


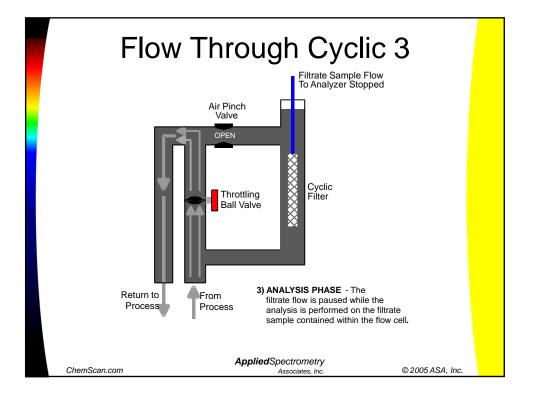


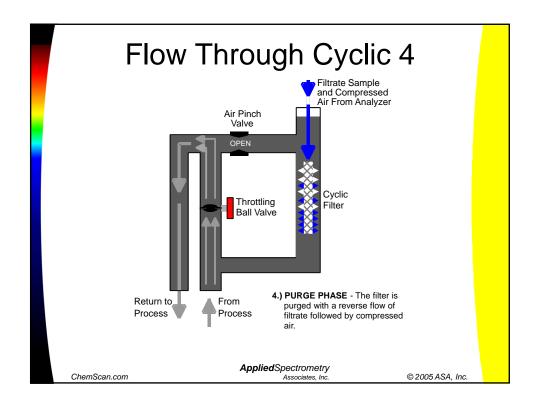


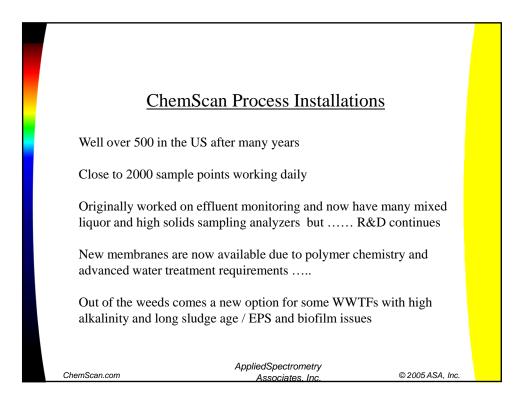


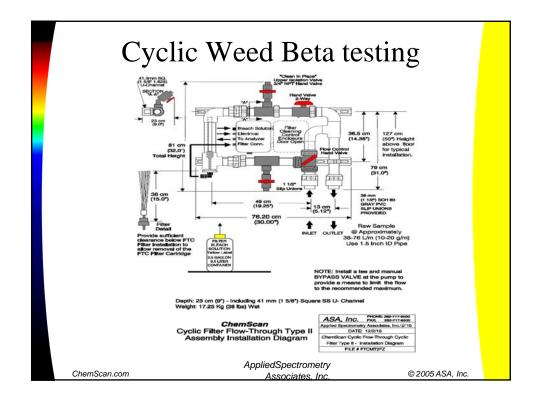




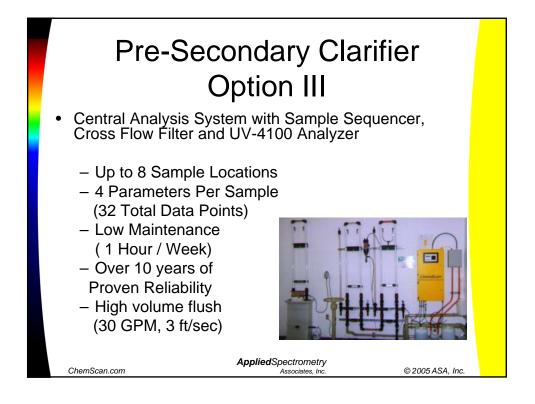


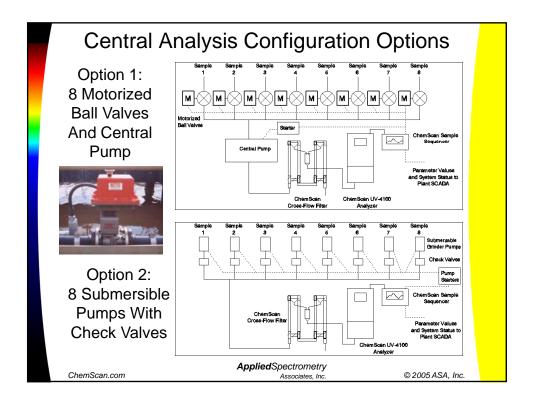


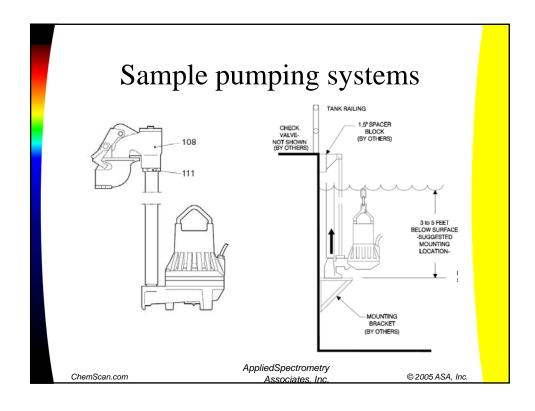


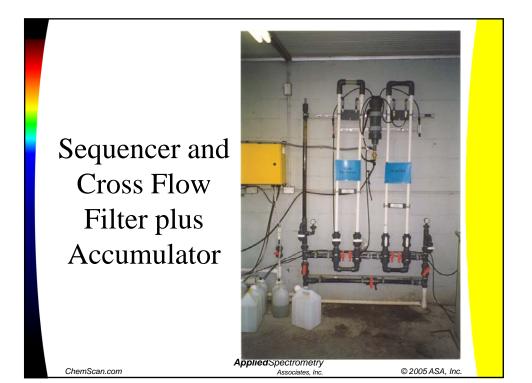


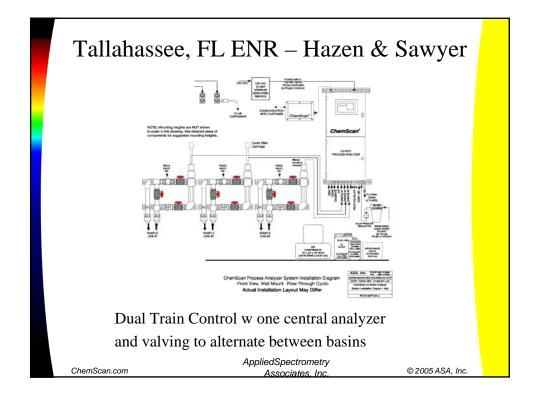


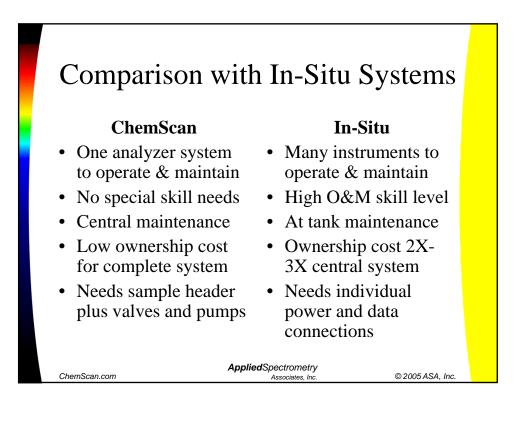


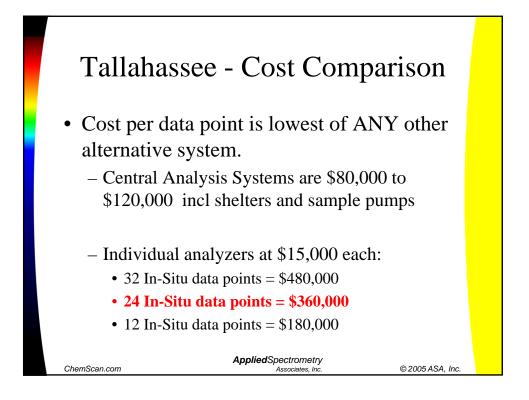


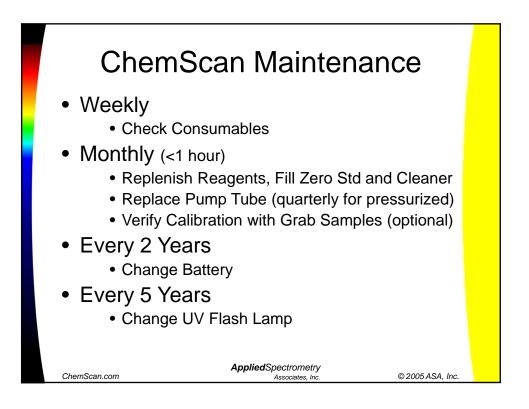














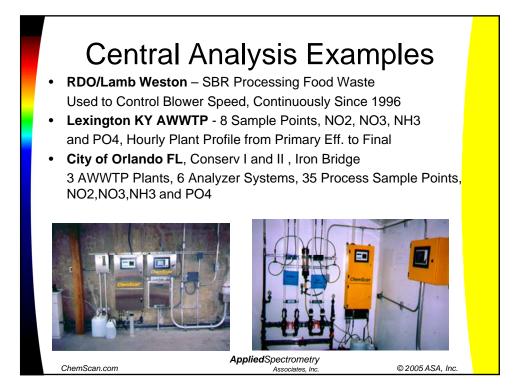
• Installation Costs

- 3-4 HP pumps are included in the system price
 - Submersible grinder pump (Flygt) is standard
- Installation includes:
 - Pump platform and guide rails
 - Pump power 230/460V and motor starters or starter panel
 - Pump control signal wire, local on/off switch
 - 2 inch PVC sample pipes and brackets
 - Check valves at drop legs
 - Central analyzer and controller power and data lines
- Estimate is \$25,000 to \$50,000 for Central Systems

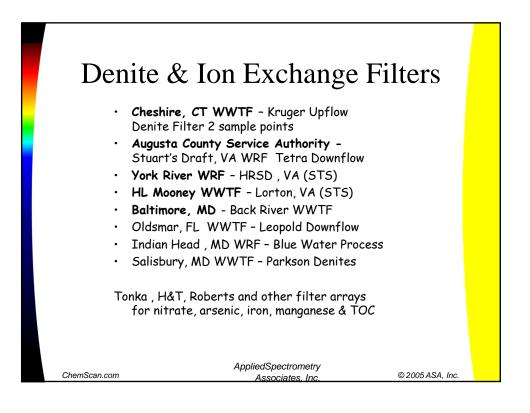
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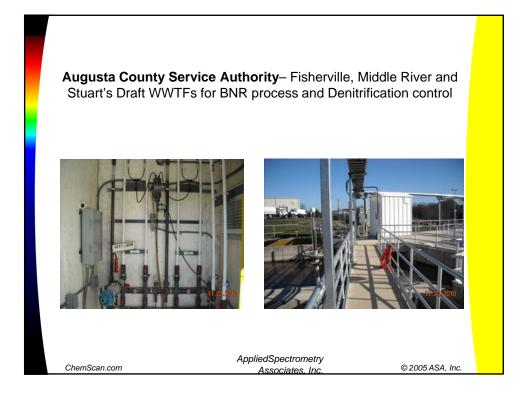
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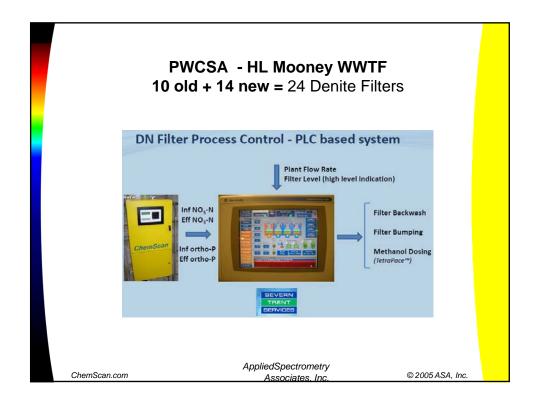
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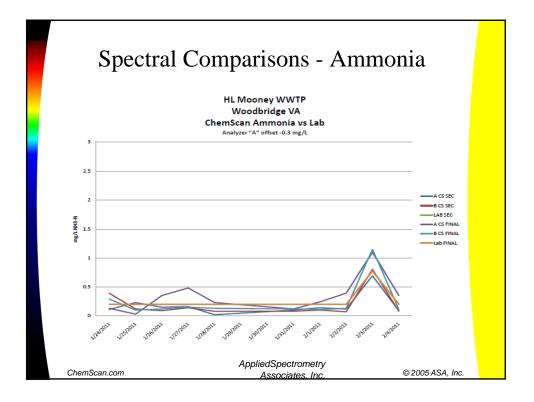


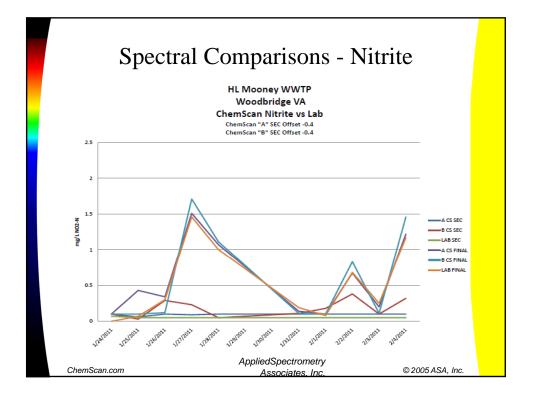
Spe	ecialty Anal	yzers
ChemScal CHLORINATION-DECH	ILORINATION ANALYZER	V-2150/DC
ChemScar WASTEWATER AMM	7° MONIA AND NITRATE	UV-2150/N
ChemScar PROCESS ANAL		UV-3150
ChemScal CHLORAMINATIO		UV-2150/S
ChemScan.com	AppliedSpectrometr Associates, Inc	

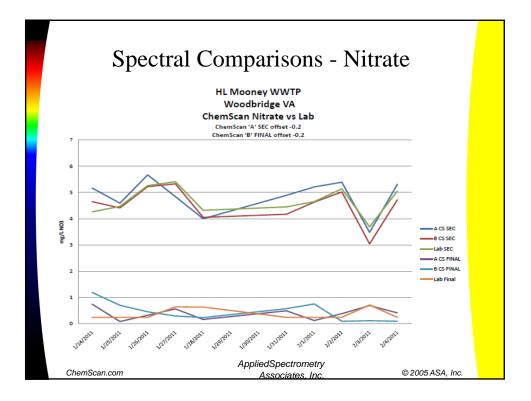


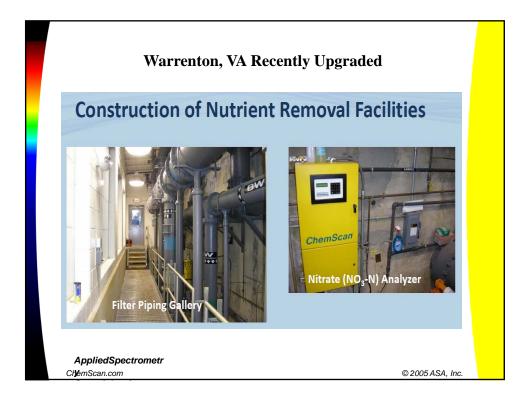


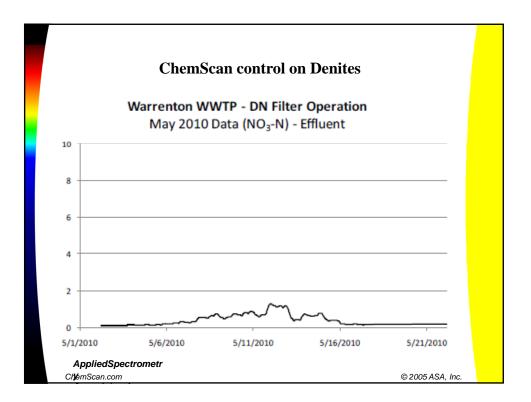


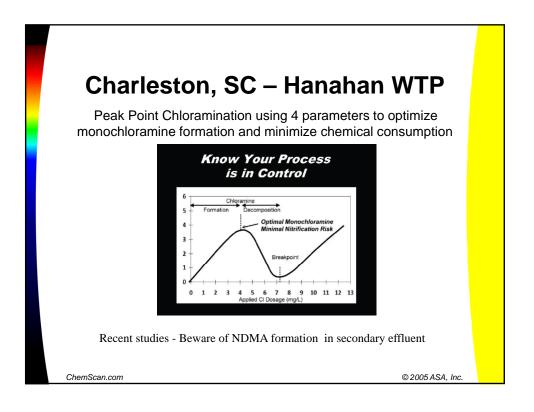


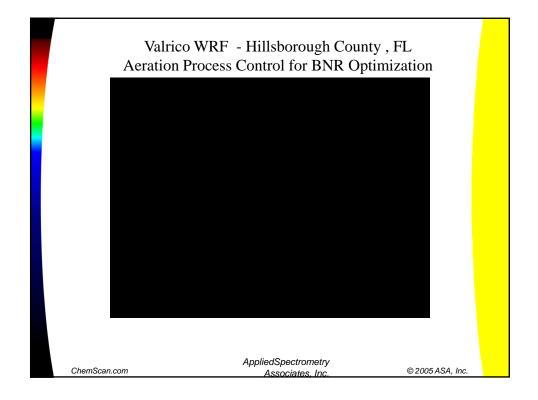


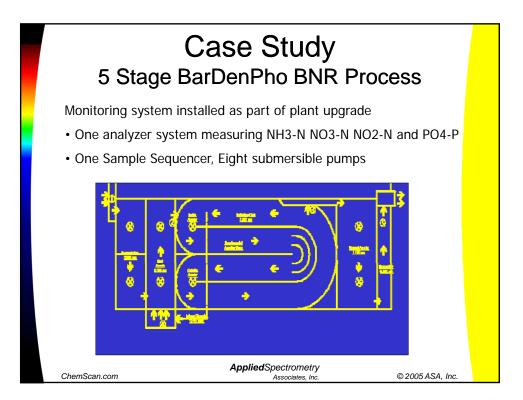


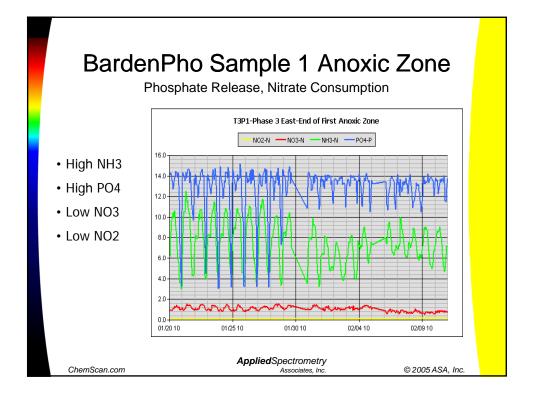


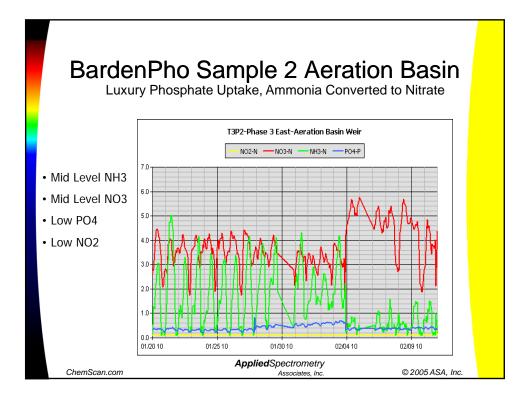


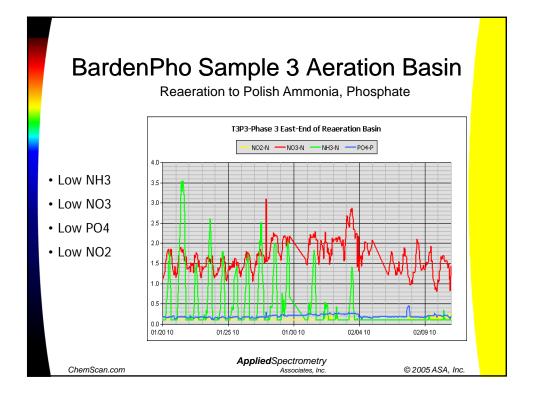


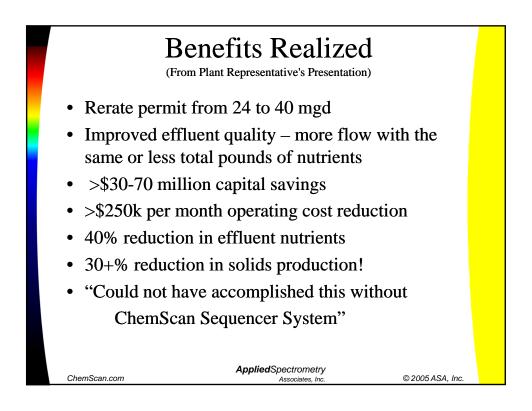


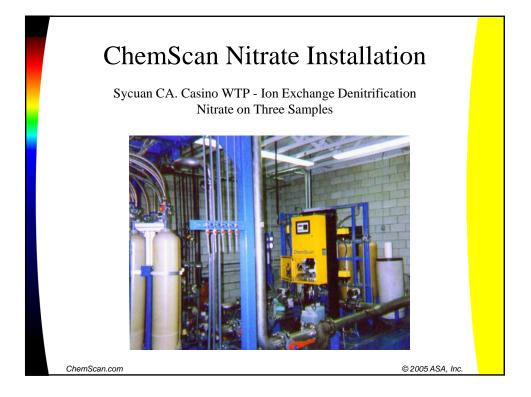


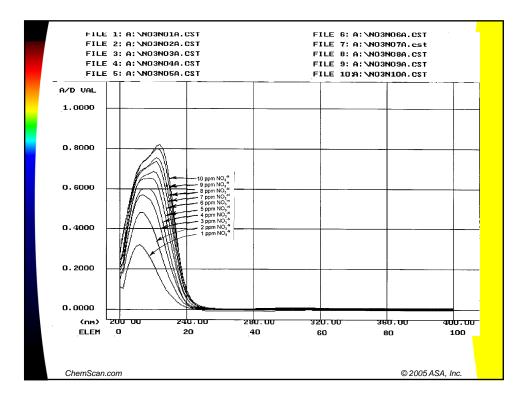


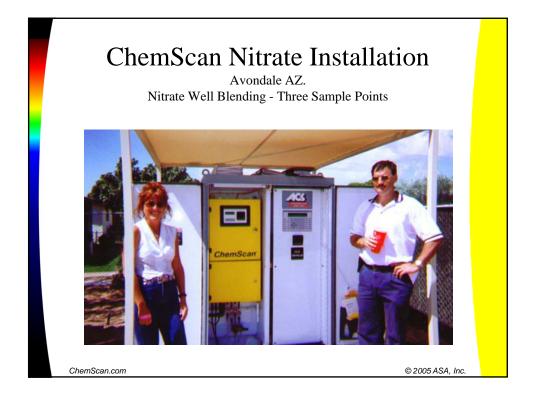


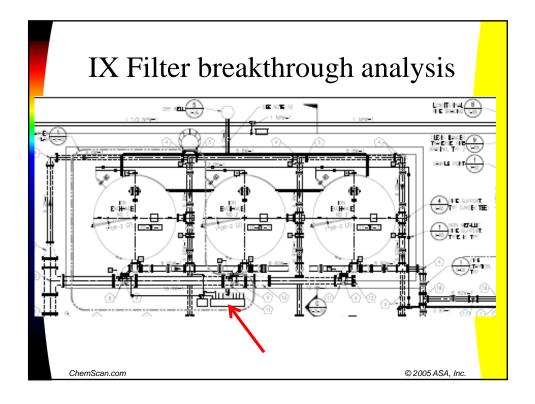




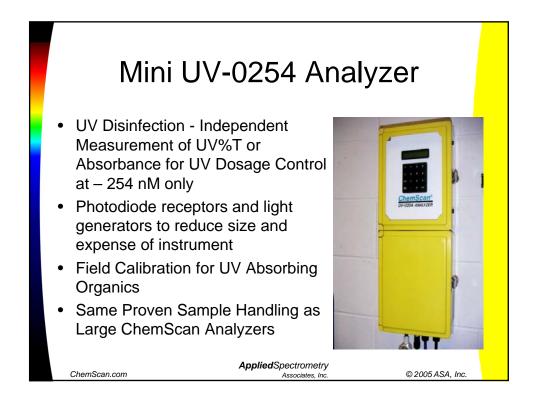


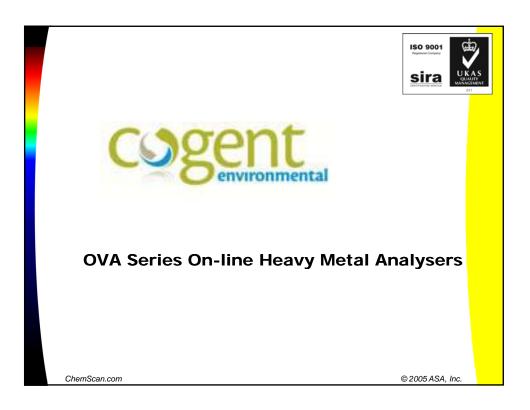


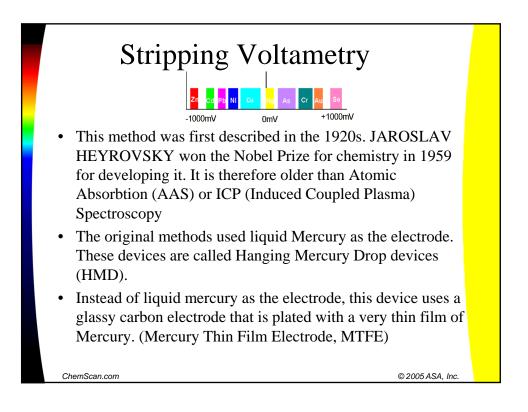












Applied Stripping Voltametry – ASV

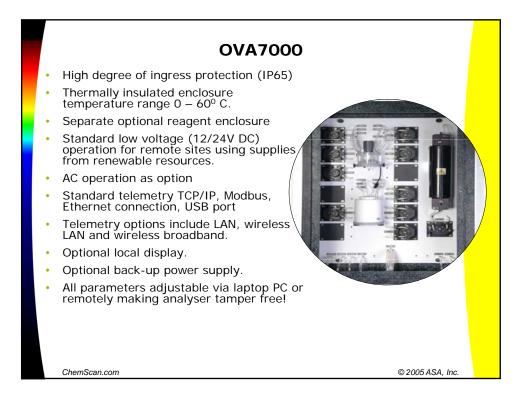
- ASV is an analytical technique that specifically detects heavy metals such as Arsenic, Cadmium, Lead, Mercury and others.
- ASV essentially works by electroplating certain metals in solution onto an electrode. This concentrates the metal. The metals on the electrode are then sequentially stripped off, which generates a current that can be measured.
- The current (milliamps) is proportional to the amount of metal being stripped off. The potential (voltage in millivolts) at which the metal is stripped off is characteristic for each metal. This means the metal can be identified as well as quantified.

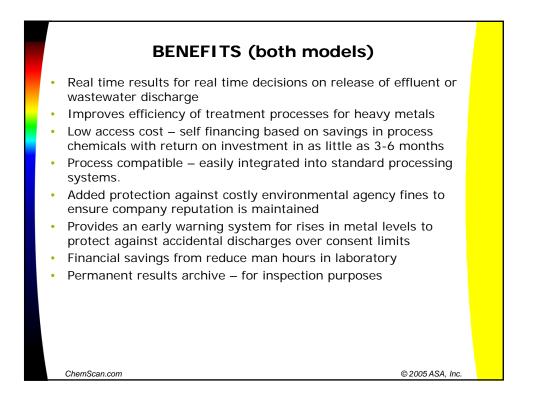
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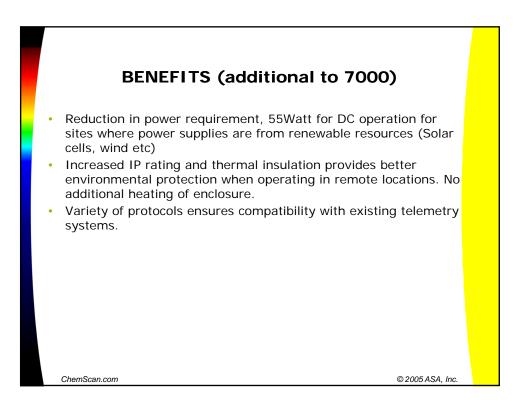
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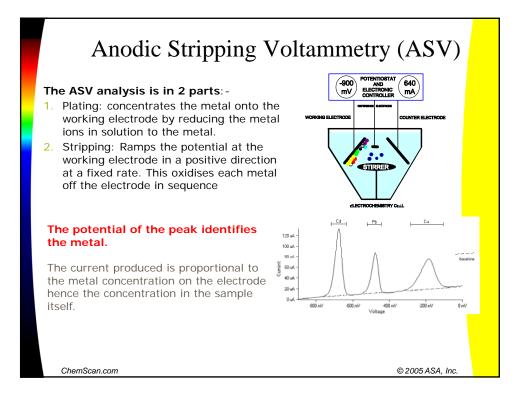
Online monitoring of heavy metals in water, river water, wastewater and industrial effluent. Real time results Configurable to customer requirements Pre-treatment options consist of acid/UV digest for elimination of potential interferences Easy to use - trained chemists are not required Applications include: As, Hg, Cr, Zn, Cd, Pb, Cu, Ni, Tl Quick and accurate analysis Excellent correlation with the standard laboratory methods (+/- 10%) ChemScan.com © 2005 ASA. Inc.

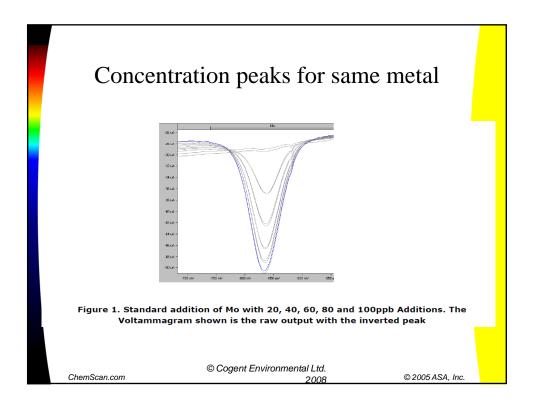
Metal	Metal Name	PDV(Portable analyser)	OVA(on-line analyser)
Ag	Silver	2µg/l	5µg/l
As	Arsenic	0.5µg/l	2µg/l
Au	Gold	5µg/l	5µg/l
Cd	Cadmium	0.5µg/l	0.5µg/l
Co	Cobalt	5µg/l	5µg/l
Cr	Chromium	5µg/l	5µg/l
Cu	Copper	1µg/l	1µg/l
Fe	Iron	5µg/l	10µg/l
Hg	Mercury	0.1µg/l	0.5µg/l
Mn	Manganese	2µg/l	5µg/l
Ni	Nickel	0.2µg/l	0.5µg/l
РЬ	Lead	0.5µg/l	1µg/l
Pd	Palladium	5µg/l	1µg/l
SЬ	Antimony	1µg/l	1µg/l
Se	Selenium	20µg/l	20µg/l
Sn	Tin	5µg/l	5µg/l
Te	Tellurium	5µg/l	5µg/l
TI	Thallium	5µg/l	5µg/l
U	Uranium	5µg/l	5µg/l
	s vary with sample type. Typical values ar for the determination of metals using ve	e shown. oltammetry from USEPA, NIOSH, ASTM, D	IN, AOAC and ISO.

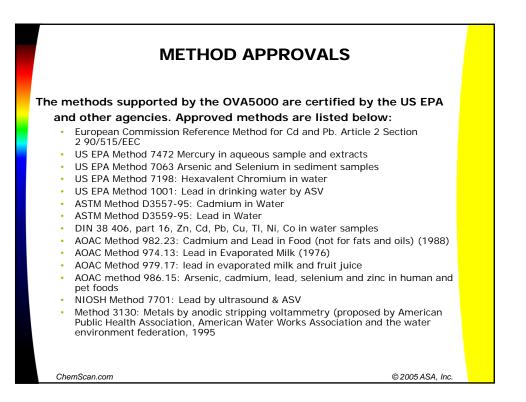


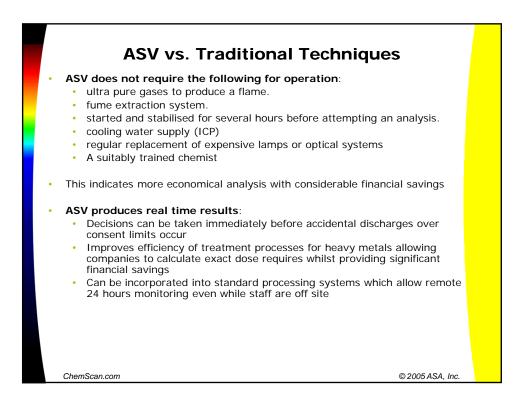


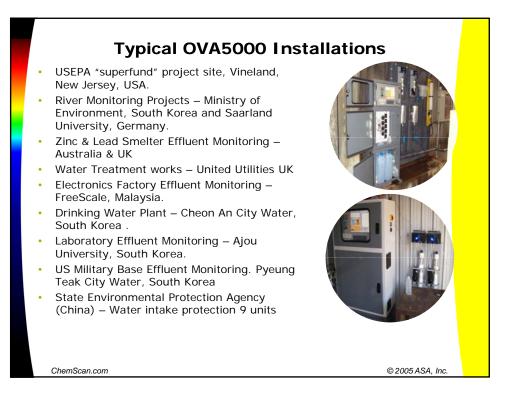


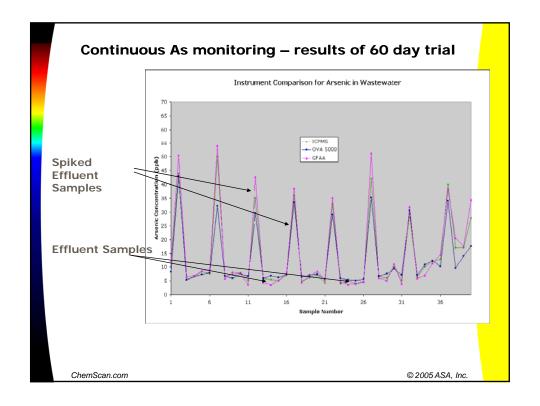




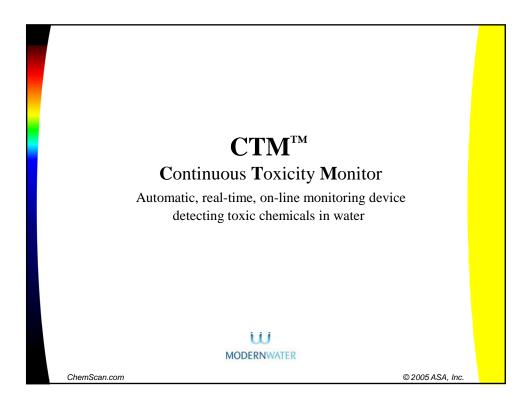








RESULTS
OVA5000 was the only instrument to complete the 60 day trial with no failures
 At high concentrations OVA5000 showed a negative bias and the GFAA a high bias compared to the reference ICPMS
 At low concentrations (<10ppb) the OVA5000 showed a correlation coefficient closer to ICPMS result than the GFAA
 Following the evaluation the OVA5000 was subsequently purchased by Sevenson Environmental Services and will be involved in the cleanup operation for the next 30 years
 Connected to plant control system so that arsenic levels in discharge are monitored 24/7
 Effluent proactively controlled to minimize chance of accidental release in to surrounding area
Offsite operators can begin corrective action if levels in effluent approaches critical level before on call personal arrive on site outside of normal working hours
Significant cost savings by closing the laboratory at weekends
 Increased protection to local people and environmental health UPDATE JUNE 2010 – Unit still operating within specification after over 4 years
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Bacterial Bioluminescence: *Vibrio*

Vibrio fischeri - bacteria sourced from deep sea fish, can be planktonic or symbiotic

Bacteria emit light when healthy, light output reduces **rapidly** on contact with toxins

'Inhibition' = the light goes out

Indicator for human toxicity : 'Biosensor' for monitoring chemical toxicity in water

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Biosensors in toxicity monitoring

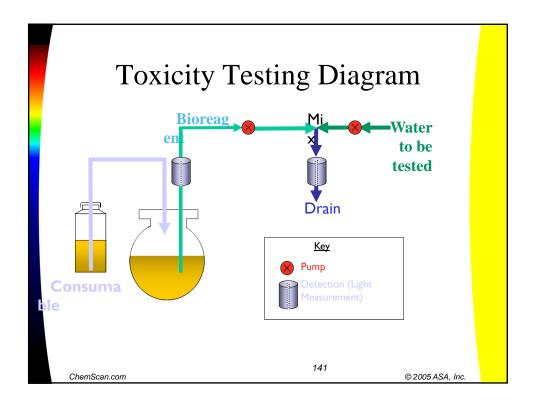
- Lab test since 1980's
- Online toxicity monitor new opportunities
 Regulatory and legislative change driving new markets
 - Post 9/11 early warning systems required
- Various on-line analysers developed but reliable longterm automatic operation and rapid detection remain elusive
- Cymtox CTMTM Continuous Toxicity Monitor
 - True early warning system modern day 'canary in a coal mine'

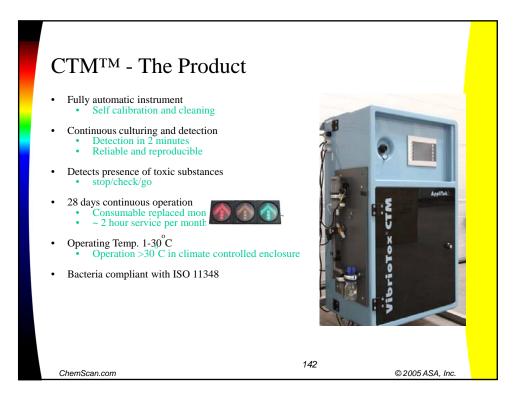




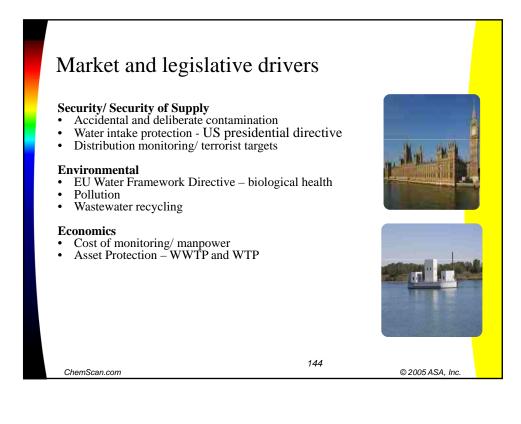
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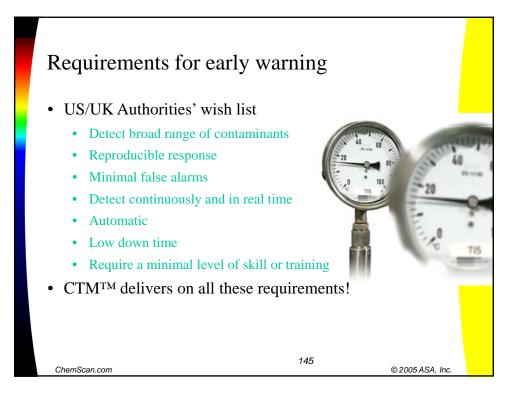
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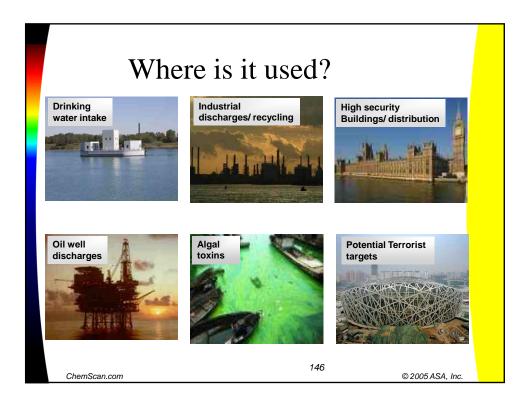




Toxin	Concentrati	Detection Level		ך ר
	on (ppm)	2 min	7 min	1
DMSO	10,000	1	2	1
Mercury Chloride	0.1	2	4	
Potassium Cyanide	2.5	3	2	1: No detection 2 : <10% inhibition
Glyphosate	0.5	3	3	
Atrazine	0.1	3	3	- 3 : 10-25% inhibiti <mark>o</mark>
Phenol	20	3	3	4: 25-50% inhibition
Zinc Sulphate	5	3	4	5:>50% inhibition.
Sodium Nitrite	0.1	3	4	
Sodium Arsenite	0.1	3	5	7
Toluene	0.1	4	3	
Malathion	0.1	4	4]
Potassium Dichromate	0.1	4	4	
Chromium (IV) Oxide	25	5	2	







	D 011	
HydroGuard Product Profile		
Any Combination of Measurements in a <u>S</u>	lingle Unit!	
✓ Free Chlorine (using DPD)		
✓ Total Chlorine (using DPD)		
✓ Free Chlorine (using amperome	trics)	
✓ Redox (ORP)		
✓ PH		
✓ Temperature		
✓ Turbidity		
✓ Conductivity		
✓ Flow		
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